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Thermodynamics?

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The first law of
thermodynamics states

that, as a system

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Undergoes a change of state, energy may cross the boundary as either heat or work, and each may be positive or negative. The net change in the energy of the system will be equal to the net energy that crosses the boundary of the system, which may change in the form of internal energy, kinetic energy, or potential

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System-Collection of
matter within prescribed
and identifiable
boundaries.

~~Tarik Al Shemmeri~~

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sounds intimidating, and
it can be. However, if
you hone in on the most
important
thermodynamic
formulas and equations,
get comfortable

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Converting from one unit of physical measurement to another, and become familiar with the physical constants related to thermodynamics, you'll be at the head of the class.

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