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Solution: $\dot{Q} = n \dot{q} = 100 \times 400 = 40000 \text{ kJ/h} = 666.7 \text{ kJ/min}$
 $dE_{\text{air}} dt = \dot{Q} = m_{\text{air}} c_v dT_{\text{air}} dt$
 $m_{\text{air}} = PV/RT = 101 \times 1500 / (0.287 \times 300) = 1759.6 \text{ kg}$
 $dT_{\text{air}} dt = \dot{Q} / m_{\text{air}} c_v = 666.7 / (1759.6 \times 0.717) = 0.53^\circ\text{C/min}$
5.119 A piston/cylinder of cross sectional area 0.01 m² maintains constant pressure.

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This introductory book on thermodynamics contains many chapters devoted to the basic principles such as defining the state and the definition and determination of work. Later chapters include cycles, combustion, equilibrium and compressible flow.

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