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3. b 2. a. alcohol b. water c. the gels 3.

The mixture is a colloid. The properties are
consistent with those reported in Table 3 on

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page 404 of the text. The particle size is small, but not too small, and the mixture

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CHAPTER 12 REVIEW Solutions SECTION 1 SHORT ANSWER Answer the following questions in the
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space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2.

Identify the solvent in each of the following examples:

mc06se cFMsr i-vi

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PLAY. 1. ... 2. A solution is a homogeneous mixture of two or more substances. A solution has at least two components. The majority component is usually called the solvent, and the minority component is usually called the

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solute.

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define an alloy define & identify
electrolytes & nonelectrolytes factors that
affect the rate of dissolution definitions of
unsaturated, saturated & supersaturated
solutions general rules for predicting
whether a solute is soluble in a solvent
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Modern Chemistry Chapter 12-Solutions Section
1- Types of Mixtures Solutions are
homogeneous mixtures of two or more
substances in a single phase. Soluble
describes a substance as capable of being
dissolved. Solvent is the dissolving medium
in a solution. Solute is the substance that
is dissolved in a solution.

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